# Development of a Stability Indicating UPLC Method for the Determination of Tirbanibulin in Bulk and its Pharmaceutical Dosage Form

## Short Title: Stability Indicating UPLC Method for Tirbanibulin

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## ABSTRACT

**Objectives:** The primary goal of this study was to create and validate a simple, precise, sensitive, and accurate UPLC method for estimating tirbanibulin in pure and dosage form.

**Materials and Methods:** The UPLC technique was developed using a Waters Acquity UPLC Phenyl (100x2.1mm,1.7m) column. The developed technique was validated in accordance with the International Conference on Harmonization's guidelines (ICH).

**Results:** Tirbanibulin was separated chromatographically with high resolutions using the mobile phase acetonitrile: buffer (30:70 v/v) at 0.5 mL/min, 5 µL injection volume, and 220 nm wavelength. The validated technique was found to be linear in the 1—15 µg/mL range. The detection and quantification limits for tirbanibulin were determined to be 0.03 and 0.1µ g/mL, respectively. The percentage RSD was less than 2%, demonstrating the precision of the developed technique. Furthermore, the recovery rate was nearly 100%, confirming the method's accuracy. Minor modifications to the chromatographic conditions demonstrated the method's robustness.

**Conclusion:** The developed analytical method was precise, simple, reproducible, and sensitive. As a result, it can be used to determine tirbanibulin.

Key words: UPLC, tirbanibulin, apoptosis, actinic keratosis, forced degradation studies

# INTRODUCTION

Actinic keratosis (AK) is a precancerous skin condition affecting the face, balding scalp, and extremities. It is caused by the proliferation of atypical keratinocytes in response to prolonged intermittent UV light exposure. AK starts with DNA damage and mutation, then progresses to neoplastic transformation and growth <sup>1</sup>. These lesions may develop into squamous cell carcinoma (SCC) once abnormal cell invasion affects the dermis structures and has the potential to metastasize <sup>2</sup>. Long-term UV exposure causes a variety of epigenetic and

genetic alterations, disrupting the activity of crucial genes in keratinocytes that promote the progression of AK to SCC <sup>3</sup>.

The chemical name of tirbanibulin is N-benzyl-2-(5-(4-(2-morpholinoethoxy) phenyl) pyridine-2-yal) acetamide. Tirbanibulin is a microtubule inhibitor. Tirbanibulin is a non-ATP-competitive inhibitor that also disrupts the proto-oncogenic Src tyrosine kinase signalling pathway<sup>4</sup>. Tirbanibulin also promotes the G2/M arrest of proliferating cell populations, upregulates p53, and triggers apoptosis by activating caspase-3 and cleaving poly (ADP-ribose) polymerase. Only topical use of tirbanibulin is currently approved. Tibranibulin should not be applied in close proximity to the mouth, lips, as well as eyes. Patients must take extra care to avoid getting tirbanibulin in their eyes or periocular region because it can cause unfavourable ophthalmic reactions. At the moment, only AKs on the face and scalp can be treated with tirbanibulin<sup>5</sup>. A thorough review of the literature revealed that there was no reported ultra-performance liquid chromatography method for estimating tirbanibulin that could indicate stability. Due to the high cost and fragility of analytical studies conducted using GC/MS or LC-MS in comparison to ultra-performance liquid chromatography (UPLC). the main focus was on developing an analytical method that was quick, precise, repeatable, and affordable. UPLC method was chosen for the developing stability-indicating method for determining tirbanibulin. According to the Q2 (R1) guidelines of the International Conference on Harmonization (ICH) procedure, the developed method was validated <sup>6</sup>. In this experiment, a UPLC method for the determination of the concentration of tirbanibulin in bulk form was established. The method was further successfully applied to the determination of tirbanibulin in pharmaceutical dosage form.

## MATERIALS AND METHODS

#### Chemicals and reagents

Tirbanibulin (Figure 1) pure bulk drug, tirbanibulin marketed dosage form (Klysiri), orthophosphoric acid, acetonitrile (HPLC grade), HPLC grade Water (Milli Q or equivalent) were the chemical materials and reagents used. All HPLC grade solvents were produced by Merck Company (Mumbai, India). The tirbanibulin drug sample was obtained from Shree Icon labs, Vijayawada, Andhra Pradesh, India as gift sample.

## Instrument and chromatographic conditions

Chromatographic analysis was accomplished by means of a Waters Acquity ultra performance liquid chromatographic system equipped with PDA detector. A Phenyl chromatographic column ( $100x2.1mm, 1.7\mu m$ ) was employed with a flow rate of 0.5 mL/min. The following additional parameters were also set: an injection volume of 5.0  $\mu$ L, and ambient column temperature. Acetonitrile and buffer were used in the mobile phase in a 30:70 ratio. There was a 2.0-minute runtime.

#### Preparation of buffer

Orthophosphoric acid of volume 1 mL was taken and dissolved in 1 liter of HPLC grade water.

#### Preparation of mobile phase

Acetonitrile and buffer were mixed in the ratio of 30:70 and filtered through 0.45µm membrane filter paper.

## Preparation of standard stock solution

Five milligrams of tirbanibulin was weighed and transferred to a 50 mL volumetric flask. It was then diluted to volume with acetonitrile (diluent) to prepare a standard solution.

# Preparation of standard working solution

Five millilitres (5 mL) of the standard stock solution were pipetted into a volumetric flask (50 mL). It was then reconstituted using the diluent (acetonitrile).

## Preparation of sample stock solution

Tirbanibulin sample of 100 mg was accurately weighed and transferred into a 10mL volumetric flask by adding some diluent (acetonitrile). Then, it was sonicated to dissolve for 15 min and diluted to volume with diluent.

#### Preparation of sample working solution

From the sample standard solution, 1 mL was pipetted out and transferred into a volumetric flask (10 mL). Then, it was made up with diluent (acetonitrile).

#### Method validation

The analytical method validation for the estimation of assay of tirbanibulin in bulk and pharmaceutical dosage form was conducted. According to ICH guidelines, the effectiveness of analytical method validation parameters such as system suitability, linearity, limit of detection (LOD), limit of quantitation (LOQ), specificity (interference and forced degradation), precision (method precision and intermediate precision), robustness and accuracy was done to show the method's suitability for the estimation of tirbanibulin assay.

## System suitability

By injecting 5  $\mu$ L of the sensitivity standard solution in 6 replicates, the system's suitability was demonstrated. Standard chromatograms were used to assess the system suitability parameters such as plate count, relative standard deviations, retention time, resolution and variation.

#### Selectivity

A comparison-based approach was employed for ensuring the absence of interference and confirming method selectivity. Chromatograms of blank samples and placebo samples were used for this purpose.

#### Linearity

Tirbanibulin concentration and peak area were plotted to assess the analytical method's linearity and demonstrate that the absorbance is directly proportional to the analyte concentration in the sample. Tirbanibulin stock solution (100  $\mu$ g/mL) was serially diluted to concentrations ranging from 1.0 to 15  $\mu$ g/mL and then tested in accordance with a test procedure. A regression line was used to calculate the correlation coefficient (CC), regression coefficient (R<sup>2</sup>), intercept, and slope.

#### Precision

To obtain method and intermediate precision, samples at 10  $\mu$ g/ml concentration level were analysed and on each validation day, individual calibrations were carried out. Six different test preparations with a strength of 10  $\mu$ g/mL each were assayed in order to evaluate the test method's precision. For intermediate precision, evaluation was completed by carrying out the required assay for six different test preparations with a strength of 10  $\mu$ g/mL. Relative standard deviation (RSD%) was used to express precision.

#### Accuracy

For tirbanibulin, a recovery study was conducted at concentrations between 50% and 150% of the initial assay value. Individual level sample solutions were made in triplicate, and test-specific method analyses were conducted. The percentages of recovery, mean recovery, and RSD were computed.

#### Sensitivity

The lower limit of quantification and the limit of detection were derived by means of the subsequent equations based upon the slope of the calibration curve and the SD of responses.

 $LOD = 3.3 \times Standard deviation (SD) / slope$ 

 $LOQ = 10 \times Standard deviation (SD) / slope$ 

## Assay of tirbanibulin

Sample (100 mg) was accurately weighed and transferred into a 10mL volumetric flask by adding some diluent. Then, it was sonicated to dissolve for 15 min and diluted to volume with diluent. From the sample standard solution, 1 mL was pipetted out and transferred into a volumetric flask (10 mL). Then, it was made up with diluent.

# Robustness

By purposefully altering chromatographic parameters, the robustness of the analytical technique was examined under a range of circumstances. The impact of the acetonitrile and buffer composition in organic phase plus (33:67) and organic phase minus (27:73), as well as impact of flow rate plus (0.55 mL/min) and flow rate minus (0.45 mL/min) were assessed during the robustness study. In all the robustness conditions, the theoretical plates, %RSD and tailing factor were evaluated as system suitability criteria. Each robustness condition received an injection of a spiked sample, and the resolution between impurities was assessed. *Forced degradation studies* 

A study was conducted to show the effective separation of degradants from tirbanibulin peak in assay method.

#### Acid degradation

Accurately weighed 1 mL of sample stock solution was transferred to a dry and clean 10 mL volumetric flask.

To this solution, 1 millilitre of 1N hydrochloric acid was added & left undisturbed for fifteen minutes. Then, 1 mL of 1N NaOH was added to neutralize and made up to the mark with diluent (acetonitrile). Samples were taken out and processed through an UPLC at certain intervals of time (0, 6, 12, 18, and 24 hours).

#### Base degradation

The possible degradation peaks and rate of degradation of tirbanibulin were assessed by weighing sample stock solution of 1 mL and transferring into a ten millilitres volumetric flask. Then, it was subjected to forced degradation by adding 1mL of 1N NaOH and left for fifteen minutes. After fifteen minutes, 1 mL of 1N HCl was added and left. Then, diluted with diluent (acetonitrile) to volume. Samples were withdrawn at specific time (0, 6, 12, 18, and 24 hrs) intervals and subjected to UPLC runs.

## Hydrolytic degradation

Tirbanibulin was tested for rate of degradation and potential degradation peaks. 1 mL of the sample stock solution was transferred to a volumetric flask with a volume of 10 mL. Then, it was forcedly degraded by adding 3 mL of HPLC grade water and kept aside for fifteen minutes. Then, sample was diluted with diluent to volume. Further, 1 mL was pipetted out into a 10 mL volumetric flask and made up with diluent (acetonitrile) up to the mark. Samples were taken out and processed through an UPLC at certain time points (0, 6, 12, 18, and 24 hours).

## Peroxide degradation

Peroxide degradation was performed by assessing the probable peaks and rate of degradation of tirbanibulin by weighing sample stock solution of 1 mL. Then, the solution was transferred into a volumetric flask of volume 10 ml. Further, it was subjected to forced degradation by adding 1mL of 10% H<sub>2</sub>O<sub>2</sub>. It was left for fifteen minutes. Then, sample was diluted to volume with diluent (acetonitrile) and mixed. Further, 1 mL was pipetted out and diluted to 10 mL with diluent. Samples were withdrawn at specific time (0, 6, 12, 18, and 24 hrs) intervals and subjected to UPLC runs.

## Reduction degradation

Tirbanibulin was tested for rate of degradation and potential degradation peaks. A volume of one millilitre of the sample stock solution was transferred into a ten millilitres volumetric flask. Then, it was forcedly degraded by adding 1 mL of 10 % sodium bi sulphate. It was undisturbed for fifteen minutes. Then, sample was diluted with diluent (acetonitrile) to volume. Samples were taken out and processed through an UPLC at certain intervals of time (0, 6, 12, 18, and 24 hours).

#### Thermal degradation

Sample of 500 mg was weighed and exposed at 105°C for 6 hrs. The exposed sample was used for analysis. The sample (100 mg) was transferred into a volumetric flask (10 mL). This sample was diluted with diluent and sonicated for 15 minutes to solubilize the contents. Further, 1 mL was pipetted into a 10 mL volumetric flask and diluted with diluent (acetonitrile). Samples at specific time (0, 6, 12, 18, and 24 hrs) points were withdrawn and put through UPLC runs to identify probable degradation chromatograms.

## Photo stability degradation

The degradation rate and possible peaks of degradation for tirbanibulin were assessed by weighing 500 mg of sample. The sample was exposed for 6 hours at sunlight. 100 mg of the above sample was weighed. Then, transferred to a volumetric flask (10 mL). Diluents were added to this and sonicated for 15 minutes to dissolve the contents. Further, from the above solution 1 mL was pipetted out into a volumetric flask of volume 10 millilitres and made up to volume with diluent (acetonitrile). Samples at specific time (0, 6, 12, 18, and 24 hrs) points were withdrawn and put through UPLC runs.

# RESULTS

#### Optimized process

After a series of trials, the mobile phase composition of acetonitrile: buffer in the proportion of 30:70 (v/v) had shown both peaks with good resolution, tailing factor, theoretical plate count. Hence this method was optimized and validated. Waters Acquity LC auto sampler enabled the elution, method development and validation of tirbanibulin. The method was proven to be simple to use, with high recovery, high sensitivity, and high specificity through thorough methodological validation. The optimized chromatogram was shown in Fig. 2.

System suitability parameters

Many analytical processes include testing for system compatibility. The system suitability characteristics were investigated and used to determine the best settings. The theoretical plate number (N) retention time, and tailing factor (T) were all studied for this purpose. Six replicate injections of the tirbanibulin standard working solution were used to carry out the procedure. It was observed that tirbanibulin retained for 0.814(average) min, with a tailing factor of not more than 1.31 in all peaks, indicating good peak symmetry. Theoretical plates were discovered to be greater than 3981 in all peaks. The results were tabulated in Table 1.

#### Linearity

The analytical method's linearity was its capacity to produce test results within a specified range that were directly proportional to the concentration of the analyte in the test sample. Seven concentration levels in the range of 1 and 15  $\mu$ g/mL were used to test the analytical method's linearity. The regression coefficient, y-intercept, and slope of regression line were calculated. The observed corelation coefficient value was 0.9996. The results were shown in Tables 2-3 and Figure 3.

#### Accuracy

A method's accuracy reflects how closely the outcomes it produces match the actual value. According to the accuracy results, % RSD was 0.11%, and percentage recovery at all three levels ranged from 99.7 to 100.9%. The results were tabulated in Table 4.

#### Precision

When multiple samples of the same homogeneous sample were taken under the specified conditions, the precision of the method was defined as the degree of agreement between the measurements obtained. Relative standard deviation was typically used to express precision. The percent relative standard deviation value for method precision results of tirbanibulin was found to be 0.47%. The percent relative standard deviation value for intermediate precision results of tirbanibulin was found to be 0.33%. The results were showed in Tables 5-6.

#### Sensitivity

The limit of quantification (LOQ) was defined as the lowest amount of analyte in a sample that can be quantitatively determined with appropriate precision, as opposed to the limit of detection (LOD), which was the lowest amount of analyte in a sample that can be detected but not necessarily quantitated. The LOD and LOQ for tirbanibulin were 0.03 and 0.1  $\mu$ g/mL respectively. The results were tabulated in Table 7.

#### Assay

According to the label claim, the drug content obtained from the values of sample solutions was found to be in the permissible range of 90-110 percent. The % assay of tirbanibulin was found to be 100.5% w/w. The results were displayed in Table 8.

#### Robustness

The influence of slight alterations in chromatographic settings was used to determine the robustness of the analytical process. The percent RSD of tirbanibulin was less than 2.0 in all of the deliberately changed chromatographic settings. The results were showed in the Table 9.

#### Selectivity

In the retention time ranges, the UPLC chromatograms for the drug matrix (combination of the medicine and placebos) revealed nearly no interference peaks. As a result, the proposed UPLC approach in this study was selective. The method's specificity and selectivity were tested by looking for interference peaks in the chromatograms of blank and placebo samples. Because of the excipients, there were no interfering peaks. As a result, the procedure was specific and selective. Figures 4 and 5 show the chromatograms of blank and working placebo solution respectively.

## Forced degradation studies

Thrbanibulin was subjected to a variety of stress conditions, which include hydrolysis, base, oxidative, acid, photo stability, and thermal degradation, as per ICH guidelines. The proposed UPLC approach was used to monitor degradation behaviour on a regular basis. The PDA detector results from the forced deterioration results revealed that the tirbanibulin peaks were pure and homogenous in all of the stressful conditions studied. This demonstrates that the approach is both particular and stable. All the results of stability studies were displayed in Tables 10 - 16. The degradation chromatograms were displayed in Figures 6- 17.

#### DISCUSSION

Tirbanibulin is a microtubule inhibitor. Tirbanibulin is a non-ATP-competitive inhibitor that also disrupts the proto-oncogenic Src tyrosine kinase signalling pathway<sup>4</sup>. Therefore, a technique for tirbanibulin determination is required. According to the ICH, a system suitability test is frequently used to assess a chromatographic system's resolution, column efficiency, and repeatability to make sure it is suitable for a certain analysis <sup>6</sup>. The new approach was tuned to produce a symmetrical peak and high theoretical plates (N). The total number of theoretical plates was above 2000, which was deemed sufficient for the system suitability test. According to the standards, the tailing factor was within the specified limitations. These findings demonstrate that the proposed strategy can produce data of acceptable quality. The suggested method's application to the analysis of formulations is its key feature. Hence, the market sample of tirbanibulin was collected and analysed by employing the proposed method. The study confirmed that the created UPLC method was accurate and easy enough to be used on a daily basis. The suggested assay method's high content results indicate that this technique can be engaged for quantitative regular quality control study of pharmaceutical dosage forms. The analytical method's linearity was its capacity to produce test findings within a specified range that directly relate to the analyte concentration in the test sample <sup>6</sup>. The regression line of analysis expresses the relationship between concentration and peak area of tirbanibulin. As a result, the findings revealed that the peak area and analyte concentration showed a strong correlation. The R<sup>2</sup> high value indicated good linearity. An analytical technique's accuracy shows the closeness of outcomes produced by that technique with the true value <sup>6</sup>. The percentage recovery and percent RSD results fell within the acceptable ranges of 98.0% to 102.0% and not more than 2.0%, respectively, demonstrating the method's suitability for routine drug analysis. The degree of agreement between a set of measurements made using repeated samples of the same homogeneous material under the specified conditions was considered the precision of the method, and it was typically stated as relative standard deviation<sup>6</sup>. The results were well under the usually accepted 2 percent limit. As a result, the new method's precision has been confirmed. The limit of quantification (LOQ) was defined as the smallest amount of analyte in a sample that could be identified quantitatively with adequate accuracy. The lowermost amount of analyte in a sample that could be detected but not necessarily quantitated was defined as the limit of detection (LOD)<sup>6</sup>. The LOD and LOQ for tirbanibulin were observed to be 0.03 and 0.1 µg/mL respectively. The system suitability parameters were not significantly changed while varying the conditions, hence the method was robust. In the retention time ranges, the UPLC chromatograms for the drug matrix (combination of the medicine and placebos) revealed nearly no interference peaks. As a result, the proposed UPLC approach in this study was selective. The drug tirbanibulin was found to undergo extreme degradation in peroxide degradation conditions. The tirbanibulin peaks were homogenous and showed results within acceptance limits in all the forced degradation studies. This demonstrates that the approach is both particular and stable. The developed method effectively passed all the validation parameters and was applied efficaciously for the determination of tirbanibulin.

## CONCLUSION

For the estimate of tirbanibulin in dosage form, a simple, accurate, and specific approach was established. Tirbanibulin had retention time of 0.811 minutes. The percent RSD of method precision and intermediate precision was found to be 0.47 and 0.33 percentage. For tirbanibulin, percent recovery was 100.1 percent. The LOD and LOQ values for tirbanibulin calculated from regression equations were  $0.03 \mu g/mL$  and  $0.1 \mu g/mL$  consecutively. Regression equation of tirbanibulin was y = 309884.02x + 29801.60. There were some degradation peaks in acid, base, thermal, reduction and peroxide stressed conditions according to the results of the forced degradation test. Because retention times and run times were reduced, the method created was simple and economic, and it might be utilized in frequent quality control tests in industries.

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## Ethics Committee Approval: Not applicable.

**Informed Consent:** Not applicable.

#### Authorship Contributions

Concept: G.P.K., S.R., Design: G.P.K., S.R., Data Collection or Processing: G.P.K., Analysis or Interpretation: G.P.K., Literature Search: G.P.K., Writing: G.P.K.

#### **CONFLICT OF INTEREST**

The authors declare that there is no conflict of competing financial interests.

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## TABLES



		1.	3024698		100	X	7	
		S. No	Area of ti	irbanibulin	% Label claim	$\bigcirc$		
le 5. Method	l precision stud	ies of tirb	anibulin					
SD. Stall	dard deviation,	, KSD. ICI	ative standa					
% RSD *SD: store	dard deviation	DSD: rol	otive standa	rd deviation	0.11			
SD					0.115			
Mean					100.1			
	1.5	1.5	13	100.1				
150%	1.5	1.4	95 02	99.7 100 1	100.2	0.61	0.6	
1 500/	1.0	1.0	07	100.7	100.0	0.61	0.6	
	1.0	0.9	97	99.7				
100%	1.0	1.0	01	100.1	100.2	0.50	0.5	
	0.50	0.5	03	100.6				
	0.50	0.4	<i>)</i>	100				

Table 5. Method precision studies of tirbanibulin

S. No	Area of tirbanibulin	% Label claim
1.	3024698	100
2.	3046952	100.7
3.	3036756	100.4
4.	3065289	101.3
5.	3026985	100.1
6.	3034356	100.3
Mean		100.5
S. D		0.476
%RSD	$2 \times$	0.47

\*SD: standard deviation, RSD: relative standard deviation

CM	S. No	Area of tirbanibulin	% Label claim
	1.	3026985	100.1
	2.	3032534	100.3
	3.	3042589	100.7
	4.	3045824	100.8
	5.	3049269	100.8
	6.	3050687	100.9
	Mean		100.6
	S. D		0.337

£ 4:..1 ...

0/DCD	
%KSD	

0.33

\*SD: standard deviation, RSD: relative standard deviation

	g	LOD (µg/mL)	LOC	) (µg/mL)	
Tirb	anibulin	0.03	0.1		
Table 8	Results of mar	keted formulation	on analysis.		
Compound nan	ne Brar	nd name	Label claim	(mg)	% Assay (%w/w)
ʻirbanibulin	Klys	siri	2.5		100.5
e 9. Robustne	ss studies of t Peak area	irbanibulin % Label cl	aim Mean	S.D	% RSI
Flow rate $(+) 0.55$	2785496	100.4	99.6	0.764	0.77
ml/min	2758463 2745818	99.4 98.9	$\mathbf{O}$		
ml/min Flow rate (-) 0.45 ml/min	2758463 2745818 3365289 3352144	99.4 98.9 100.2 99.9	100.1	0.173	0.17
ml/min Flow rate (-) 0.45 ml/min Mobile phase (+) 33	2758463 2745818 3365289 3352144 3362569 2652542 2665324 265324	99.4 98.9 100.2 99.9 100.2 99.4 99.9	100.1 99.7	0.173	0.17 0.25

\*SD: standard deviation, RSD: relative standard deviation

Table 10. Acid degradation studies of tirbanibulin

Time	Peak area	% Label claim	% Degraded	Purity angle	Purity threshold
Initial	2896528	95.8	4.2	0.157	1.052
6 hrs	2701953	89.4	10.4	0.148	1.058
12 hrs	2542200	84.1	15.7	0.136	1.049

18 hrs	2416121	79.9	19.9	0.124	1.052	
24 hrs	2318506	76.7	23.1	0.125	1.063	

Table 11. Base degradation studies of tirbanibulin

legradation	n studies of tir	banibulin			
Time	Peak area	% Label claim	% Degraded	Purity angle	Purity threshold
Initial	2926358	96.8	3.2	0.163	1.047
6 hrs	2765472	91.5	8.5	0.166	1.047
12 hrs	2556387	84.6	15.4	0.139	1.037
18 hrs	2446953	81	19	0.149	1.071
24 hrs	2384105	78.9	21.1	0.129	1.034
		5			

 Table 12. Peroxide degradation studies of tirbanibulin

C	Time	Peak area	% Label claim	% Degraded	Purity angle	Purity threshold
	Initial	2905874	96.1	3.9	0.144	1.057
	6 hrs	2763592	91.4	8.6	0.142	1.053
	12 hrs	2575219	85.2	14.8	0.159	1.066
	18 hrs	2429856	80.4	19.6	0.162	1.062
V.	24 hrs	2306598	76.3	23.7	0.168	1.069

 Table 13. Reduction degradation studies of tirbanibulin

Time	Peak area	% Label claim	% Degraded	Purity angle	Purity threshol	ld
						<b>(</b>
Initial	2910368	96.3	3.7	0.169	1.032	X
6 hrs	2758292	91.3	8.7	0.171	1.063	
12 hrs	2552875	84.5	15.5	0.129	1.042	
18 hrs	2455937	81.3	18.7	0.154	1.069	
24 hrs	2340156	77.4	22.6	0.125	1.044	SO.

Table 14. Thermal degradation studies of tirbanibulin

Time	Peak	% Label	%	Purity	Purity
	area	claim	Degraded	angle	threshold
Initial	2896237	95.8	4.2	0.135	1.064
6 hrs	2845421	94.1	5.9	0.129	1.072
12 hrs	2790124	92.3	7.7	0.145	1.076
18 hrs	2729638	90.3	9.7	0.138	1.068
24 hrs	2609856	86.3	13.7	0.147	1.064
		K //			

Table 15. Photolysis degradation studies of tirbanibulin

Time	Peak area	% Label claim	% Degraded	Purity angle	Purity threshold
Initial	2997541	99.2	0.8	0.178	1.074
6 hrs	2976412	98.5	1.5	0.149	1.069
12 hrs	2954638	97.8	2.2	0.143	1.052
18 hrs	2938721	97.2	2.8	0.146	1.054
24 hrs	2906352	96.2	3.8	0.158	1.069

		area	claim	Degraded	angle	threshold	
	Initial	2980574	98.6	1.4	0.149	1.067	
	6 hrs	2967521	98.2	1.8	0.143	1.064	
	12 hrs	2932547	97	3	0.148	1.062	
	18 hrs	2916359	96.5	3.5	0.146	1.062	
	24 hrs	2902593	96	4	0.164	1.065	
				×Q	ç		
			C				
FIGURES	0						

 Table 16. Hydrolytic degradation studies of tirbanibulin



# Figure 4. Chromatogram of blank







# Figure 10. Peroxide degradation chromatogram of tirbanibulin - Initial







